PII: S0040-4039(96)00715-0

Synthesis of 2',3'-Fused (3.3.0) γ-Butyrolactone-Nucleosides and Coupling with Amino-Nucleosides To Give Amide-Linked Nucleotide-Dimer Analogues¹

Morris J. Robins,* Sanchita Sarker, Meiqiang Xie, Weijian Zhang, and Matt A. Peterson*

Department of Chemistry and Biochemistry, Brigham Young University, Provo, Utah 84602-5700

Abstract: Stereoselective hydrogenation of Wittig products obtained readily (via the 3'-ketones) from 2',5'-bis-O-(tert-butyldimethylsilyl)nucleosides provides efficient access to 2',3'-fused γ-butyrolactone-nucleosides that can be coupled with 5'-amino-nucleosides (2-hydroxypyridine catalysis) to give amide-linked nucleotide-dimer analogues. Copyright © 1996 Elsevier Science Ltd

Interest in 2',3'-fused (3.3.0) γ-butyrolactone-nucleoside systems has been stimulated by the possibility of branched-chain nucleoside analogues functioning as antitumor and antiviral agents, the reactivity of amine nucleophiles with such strained lactones, and general interest in the development of free radical-mediated routes to fused systems. 2-5 We had considered³ that the enhanced reactivity of these lactones might allow synthesis of amide-linked nucleotide-analogues directly, without additional protection/deprotection and purification steps required for conventional peptide-bond formation with carboxylic acids and coupling reagents and/or activated esters (Scheme 1). Intensive recent effort has been expended in the preparation of oligonucleotide analogues with modified backbone structures designed to circumvent physical and biological limitations of the natural phosphodiester linkage. Among various replacements investigated, amide linkages have been shown to possess favorable properties including increased duplex stability, increased resistance to nucleolytic degradation, and enhanced membrane permeability owing to the decreased backbone charge.

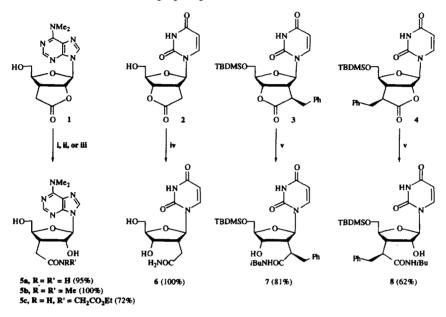
Scheme 1. Amide-Linked Oligoribonucleotide Analogues

HO
$$B_1$$
 B_2 B_2 B_2 B_3 B_4 B_5 B_5 B_6 B_7 B_8 B_8 B_9 B

Major progress in the synthesis of oligodeoxynucleotide analogues has been achieved and a variety of "antisense" oligomers have been prepared and evaluated biologically. Much less work has been reported with oligoribonucleotide analogues owing to enhanced difficulties with a 2'-hydroxyl function present. However, it has been shown that oligomers with substituents at C2' which confer the "ribo-like" rather than "deoxy-like" conformation exhibit enhanced binding with RNA, additional nuclease stability, and other favorable properties.⁸

The first γ -butyrolactone-nucleoside 1 was prepared by coupling a carbohydrate precursor with a purine.² Both 2',3'- and 3',2'-fused types (2-4) have been reported recently.³⁻⁵ The reactivity of lactone-nucleosides with ammonia and certain amines (Scheme 2) supported possible applications for the synthesis of amide-linked ribonucleotide-analogue dimers that could be incorporated into nuclease-resistant oligoribonucleotides.

Scheme 2. Prior Studies on Ring Opening of Nucleoside Lactones with Ammonia or Amines^a



^a(i) NH₃ (liq)/6 h.² (ii) HNMe₂/0 °C/4 h.² (iii) H₂NCH₂CO₂Et/DMF/30 h.² (iv) NH₃/H₂O.⁵ (v) i-BuNH₂/AlCl₂/16 h.⁴

We synthesized⁹ the γ-butyrolactone-nucleosides 15 and 16 from precursors 9¹⁰ and 10³ (Scheme 3). The readily available 2',5'-bis-*O*-(*tert*-butyldimethylsilyl)-3'-keto-(adenosine and uridine)¹¹ were treated with [(ethoxycarbonyl)methylene]triphenylphosphorane in dichloromethane at reflux to give the Wittig adducts 9¹⁰ and 10³ as single diastereomers (80-90%). No attempts were made to determine the configurations of these intermediates since stereoselective hydrogenation of 9 to 11 had been demonstrated.¹² Analogous reduction of 10 (10% Pd•C/MeOH/5 psi/3 days/ambient temperature) was less stereoselective (≥4:1) than with the adenosine analogue. Structure 12 was indicated by difference NOE experiments (6% enhancement of the 3'-proton resonance upon irradiation of the H2' signal) for the major diastereomer and this assignment was confirmed by

conversion of 12 to the lactone (Scheme 3). The 5'-O-TBDMS group was removed selectively from 9 and 10 with 90% aqueous trifluoroacetic acid for 20 min at 0 °C.¹³ Catalytic hydrogenation of 13 and 14 and treatment of the products with Bu₄NF/THF gave the γ -butyrolactone-nucleosides 15 and 16 (67 and 77%, respectively). Treatment of 11 and 12 with TBAF/THF gave 15 and 16 (72 and 85%, respectively) which confirmed stereoselective hydrogenation from the β face in all cases.

Scheme 3. Synthesis of Nucleoside Lactones

Aqueous solutions containing lactones 15 or 16 and 5'-amino-5'-deoxyadenosine (17) or 5'-amino-5'-deoxyuridine (18) at various pH values were stirred at elevated temperatures, and other solvents and reaction conditions were investigated, without success. Similar lack of reactivity of γ-butyrolactone-nucleosides 3 and 4 with isobutylamine had been noted, and aluminum(III) chloride catalysis was required to promote formation of amides.⁴ We examined several "acylation promoters" including 1-hydroxybenzotriazole (HOBT), pyrazole, 1,2,4-triazole, and 2-hydroxypyridine, and the latter proved effective. A solution of 15, 5'-amino-5'-deoxyadenosine (17, 5 equiv.), and 2-hydroxypyridine (2 equiv.) in DMF was stirred for 24 h at 70 °C to effect lactone ring opening. The amide-linked nucleotide-analogue dimer 19 was obtained in 60% yield (plus starting materials and decomposition products from the amino-nucleoside 17). Analogous treatment of 16 with 17 (5 equiv.) and 2-hydroxypyridine (2 equiv.) in DMF for 30 h at 70 °C gave 20 (82%).

Scheme 4. Synthesis of Amide-Linked Dinucleotide Analogues

In summary, readily available 2',5'-bis-O-TBDMS nucleosides have been efficiently converted into 2',3'-fused (3.3.0) γ -butyrolactone-nucleosides in gram-scale quantities. These lactones are resistant to ring opening with 5'-amino-5'-deoxynucleosides. However, they can be induced to undergo coupling to give amide-linked nucleotide-analogue dimers in good yields with an excess of the 5'-amino-5'-deoxynucleoside in the presence of 2-hydroxypyridine. A sequence of ester saponification, activation of the carboxylic acid, coupling with the amino-nucleosides, and selective deprotection/protection is under investigation to give amide dimers suitable for synthesizer-mediated incorporation into oligomers of defined sequence.

Acknowledgment: We thank the American Cancer Society (Grant DHP-34) and Brigham Young University development funds for generous support.

References and Notes

- 1. Nucleic Acid Related Compounds. 90. Part 89 is: Wnuk, S. F.; Robins, M.J. J. Am. Chem. Soc., in press.
- 2. Rosenthal, A.; Baker, D. A. J. Org. Chem. 1973, 38, 198-201.
- 3. Zhang, W. Ph.D. Dissertation, Brigham Young University, 1992.
- (a) Velázquez, S.; Huss, S.; Camarasa, M.-J. J. Chem. Soc., Chem. Commun. 1991, 1263-1265.
 (b) Velázquez, S.; Jimeno, M. L.; Huss, S.; Balzarini, J.; Camarasa, M.-J. J. Org. Chem. 1994, 59, 7661-7670.
- 5. Lawrence, A. J.; Pavey, J. B. J.; O'Neil, I. A.; Cosstick, R. Tetrahedron Lett. 1995, 36, 6341-6344.
- (a) Cohen, J. S., Ed. Oligodeoxynucleotides. Antisense Inhibitors of Gene Expression; CRC Press:
 Boca Raton, 1989.
 (b) Uhlmann, E.; Peyman, A. Chem. Rev. 1990, 90, 544-584.
 (c) Crooke, S. T.;
 Lebleu, B. Eds. Antisense Research and Applications; CRC Press: Boca Raton, 1993.
 (d) Varma, R. S. Synlett 1993, 621-637.
 (e) Milligan, J. F.; Matteucci, M. D.; Martin, J. C. J. Med. Chem. 1993, 36, 1923-1937.
- (a) De Mesmaeker, A.; Lebreton, J.; Waldner, A.; Fritsch, V.; Wolf, R. M.; Freier, S. M. Synlett 1993, 733-736.
 (b) Idziak, I.; Just, G.; Damha, M. J.; Giannaris, P. A. Tetrahedron Lett. 1993, 34, 5417-5420.
 (c) De Mesmaeker, A.; Waldner, A.; Fritsch, V.; Lebreton, J.; Wolf, R. M. Bull. Soc. Chim. Belg. 1994, 103, 705-717.
- 8. See: Cummins, L. L.; Owens, S. R.; Risen, L. M.; Lesnik, E. A.; Freier, S. M.; McGee, D.; Guinosso, C. J.; Cook, P. D. *Nucleic Acids Res.* 1995, 23, 2019-2024; and references quoted therein.
- 9. All new compounds were characterized by ¹H and ¹³C NMR spectroscopy and had elemental alalyses values within ±0.4% of theory and/or HRMS values within ±6 ppm of theory.
- 10. Usui, H.; Ueda, T. Chem. Pharm. Bull. 1986, 34, 15-23.

- 11. Hansske, F.; Madej, D.; Robins, M. J. Tetrahedron, 1984, 40, 125-135.
- 12. Lee, K.; Wiemer, D. F. J. Org. Chem. 1993, 58, 7808-7812.
- 13. Robins, M. J.; Samano, V.; Johnson, M. D. J. Org. Chem. 1990, 55, 410-412.
- (a) Beyerman, H. C.; van den Brink, W. M. Proc. Chem. Soc. 1963, 266. (b) Openshaw, H. T.;
 Whittaker, N. J. Chem. Soc. (C) 1969, 89-91.

(Received in USA 27 February 1996; revised 1 April 1996; accepted 2 April 1996)